Salt Lake Community College, Chemistry Department

Chem 1110 Workshop 1

Topic: Number, Units and Conversions

Objective:

- Review some simple math rules
- Physical Quantities: Units and Scientific Notation
- How to do Dimensional Analysis?
- Density

Simple Math Rules Review:

 Multiplication Rule (add the exponents): a¹ x a¹ = a², a^m x aⁿ = a^{m+n}

 $a \times b = b \times a$ (order is not important)

• Division (subtract the exponents): $a^4/a^2 = (a^{4-2}) = a^2$

 $a^{m}/a^{n} = a^{m-n}$

Prefixes used in the SI unit (Metric system):

giga	G	10 ⁹
mega	Μ	10 ⁶
kilo	k	10 ³
deci	d	10 ⁻¹
centi	С	10 ⁻²
milli	m	10 ⁻³
micro	u	10-6
nano	n	10 ⁻⁹

Example= 1 (Gm) gigameter = 1 x10⁹ m

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1 megameter= 1 x10<sup>6</sup> m
1 cm = 1 x 10<sup>-2</sup> m or 1/100 m or 0.01 m
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Note: In SI system all lengths units will be based on meter (m).

How to do Dimensional Analysis?

For calculations to yield the correct answers they MUST also yield the correct units.

- The best way to do dimensional analysis is to use CONVERSION FACTOR(S).
- Selection of a conversion factor:



Density:

The physical property that relates the mass of an object to its volume, mass per unit volume

density = mass (g)

volume (cm³ or mL)

Volume Unit: 1 liter, L

What is it?

 $1 L = 1000 cm^3$ (by definition)

But... using prefix milli, milliliter (mL) is... 1 L = 1000 mL

therefore, $1 \text{ mL} = 1 \text{ cm}^3$

Practice Problems:

- 1. Do the following calculations (final answer in scientific notation):
 - a) (6×10⁴⁰⁰) x (3×10⁷⁰⁰)
 - b) (48 x 10¹²⁰⁰)/(2 x 10⁻⁴⁰⁰)
- 2. Convert 120 pounds (lb) to grams (g). Note: 1 lb = 453.6 g
- 3. Convert 120 milligrams (mg) to kilograms (kg).

4. Convert the speed of 515 meters per second (m/s) to miles per hour (mi/hr).

5. The recommended adult dose of Ceftriaxone, an antibiotic, is 10.00 mg/kg of body

mass. Calculate the dose in milligrams for a 150-lb adult. 1 lb = 453.59 g.

- 6. An extra-strength aspirin contains 0.500 g of aspirin. How many grains is this? (1 grain = 64.8 mg)
- 7. An object has a mass of 40.1 g and occupies a volume of 9.51 mL. The density of this object is...